# Spin-orbit and spectroscopies from atoms to solids



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#### Plan of the lectures

#### Lecture 1

- Generalities on SOI
- Techniques: XAS, PES, RIXS
- Polarized electrons: XMCD and SP-PES
- Exemples

#### Lecture 2

- The Rashba effect
- From Rasba to TI's
- SOI and RIXS

#### **General consideration**



#### Spin-orbit: where from?



#### Spin-orbit: hydrogen atom

A magnetic field appears in the electron's rest frame:

$$\vec{B} = -\frac{1}{c^2} \vec{v} \times \vec{E} = \frac{1}{m_e c^2} \vec{p} \times \frac{1}{q} \frac{dV(r)}{dr} \frac{\vec{r}}{r} = -\frac{1}{qm_e c^2} \vec{r} \times \vec{p} \frac{1}{r} \frac{dV(r)}{dr} = \frac{1}{qm_e c^2} \frac{1}{r} \frac{dV(r)}{dr} \vec{L} ,$$

Which entails the interaction term (Thomas' s precession yields the extra 1/2):

$$E_{SO} = -\frac{1}{2} \vec{M} \cdot \vec{B} = -\frac{q}{m_e} \vec{S} \cdot \vec{B} \quad \rightarrow \quad H_{SO} = \frac{1}{m_e^2 c^2} \frac{1}{r} \frac{dV(r)}{dr} \vec{L} \cdot \vec{S}$$
$$V = -\frac{Ze^2}{r}; \quad \alpha = \frac{e^2}{\hbar c} = \frac{1}{137}; \qquad H_{SO} = \frac{Z\hbar^2 \alpha^2}{2m_e^2 e^2} \frac{1}{r^3} \vec{L} \cdot \vec{S} = \boldsymbol{\xi}(r) \vec{L} \cdot \vec{S}$$
$$\vec{L} \cdot \vec{S} = \frac{1}{2} (J^2 - L^2 - S^2)$$

#### Spin-orbit: hydrogen(oid) atom (Z protons, 1 electron)

Without spin-orbit: 
$$\mathbf{L}^2$$
;  $\mathbf{L}_z$ ;  $\mathbf{S}^2$ ;  $\mathbf{S}_z$ ;  $\mathbf{J}^2$ ;  $\mathbf{J}_z$  commute. Eigenstates:  
 $|n;l;m_l;\pm\rangle$ ;  $j = l \pm \frac{1}{2}$ 

With spin-orbit:  $\mathbf{H}_{so}$ ;  $\mathbf{J}_z$ ;  $\mathbf{J}^2$  commute. The eigenstates  $|n; j; m_j \rangle$ are appropriate combinations of the  $|n; l; m_l; \pm \rangle$ 

$$E_{nlj} = \langle jm_j | H_{SO} | jm_j \rangle = \frac{Z\hbar^2 \alpha^2}{2m_e^2} \langle \frac{1}{r^3} \rangle \langle jm_j | \vec{S} \cdot \vec{L} | jm_j \rangle = \begin{cases} \mathbf{angular} \\ \mathbf{$$

#### Textbook example: the 2p states of hydrogen

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**VECTOR MODEL** 



#### **Preliminary considerations**

The SOI couples the orientation of the magnetic moments to the crystal directions

 $\rightarrow$  magnetic anisotropy, new quantum states...

Moreover, e.m. waves interact with L, not with S. However SOI couples L and S, so that we can probe S with light!

- → Circular magnetic dichroism: absorption (XMCD); photoemission (SP-PES)
- → Resonant inelastic x-ray scattering

### Spectroscopic probes



#### PES – ARPES – SP-ARPES







#### The Mott polarimeter – spin-orbit, again !



#### XAS: unoccupied states



#### Beyond the single particle picture

## The XAS~DOS picture breaks down when the final states are strongly correlated: TM 3d states



The atomic-like multiplet structure is a fingerprint of the *ground state* configuration...

#### Photons do it all: resonant inelastic x-ray spectroscopy (RIXS)



Example - Cu L3 RIXS:  $3d^9 \rightarrow 2p^5 3d^{10} \rightarrow 3d^9$ 



#### Spin-orbit: rare gases





#### Solids: core levels - PES

#### HANDBOOK OF X-RAY PHOTOELECTRON SPECTROSCOPY



#### Solids: core levels - PES



De Groot and Kontani (2008)

#### Solids: core levels - XAS



#### Solids: valence bands - PES

2 Li

#### **PES – noble metals**







#### Solids: valence bands - ARPES



#### The Fano effect: polarized electrons from unpolarized atoms



Goal(s): to determine separately the spin- and orbital moments (and the expectation values of other observables)

**XAS**: Transitions into low-lying (localized) correlated final states, polarized by an external field or because of magnetic order

- The Pauli principle enforces spin anisotropy

**PES**: Transitions to free-electron final states, weakly correlated and non-polarized

- No need for macroscopic magnetization
- typically needs spin detection

#### Dichroism in XAS



An atomic model

Stöhr and Siegmann

#### XMCD sum rules: quantitative determination of m<sub>orb</sub> e m<sub>spin</sub>



#### Photoemission – Polarization and spin detection

TABLE I. The six fundamental photoemission spectra  $I^{xy}$ , which are linear combinations of the primitive spin-polarized spectra. x = 0, 1, 2 denotes isotropically, circularly, and linearly polarized radiation, respectively. y = 0, 1 denotes without and with spin-polarization measurement, respectively. In the column labeled z, a value 1, 2, or 3 denotes that  $\langle M \rangle$ ,  $\langle M^2 - \frac{1}{3}J(J + 1) \rangle$ , or  $\langle M^3 - \frac{3}{5}M[J(J+1) - \frac{1}{3}] \rangle$  in the ground state has to be nonzero to obtain the spectrum; z = 0 denotes the value of the monopole, which is always unity.

Ixy	Combination of primitive spectra	Z	Significance
I <sup>00</sup>	$I_{1\uparrow} + I_{0\uparrow} + I_{-1\uparrow} + I_{1\downarrow} + I_{0\downarrow} + I_{-1\downarrow}$	0	Isotropic spectrum
$I^{01}$	$I_{1\uparrow} + I_{0\uparrow} + I_{-1\uparrow} - I_{1\downarrow} - I_{0\downarrow} - I_{-1\downarrow}$	1	Spin spectrum
$I^{10}$	$I_{11} - I_{-11} + I_{11} - I_{-11}$	1	Orbit spectrum (MCD)
$I^{11}$	$I_{1\uparrow} - I_{-1\uparrow} - I_{1\downarrow} + I_{-1\downarrow}$	0,2	Spin-orbit spectrum
I <sup>20</sup>	$I_{11} - 2I_{01} + I_{-11} + I_{11} - 2I_{01} + I_{-11}$	2	Anisotropic spectrum (MLD)
<i>I</i> <sup>21</sup>	$I_{1\uparrow} - 2I_{0\uparrow} + I_{-1\uparrow} - I_{1\downarrow} + 2I_{0\downarrow} - I_{-1\downarrow}$	1,3	Anisotropic spin magnetic spectrum

#### Core level photoemission



#### Valence band: Integrated spin-orbit spectrum

$$\frac{\boldsymbol{\rho}^{11}}{\boldsymbol{\rho}^{00}} = \frac{2\boldsymbol{A}_1}{\boldsymbol{A}_0} \frac{\left\langle \sum_i \boldsymbol{l}_{zi} \, \boldsymbol{s}_{zi} \right\rangle}{\langle \boldsymbol{n} \rangle} = \frac{2\boldsymbol{A}_1}{\boldsymbol{A}_0} \frac{\left\langle \sum_i \boldsymbol{l}_i \cdot \boldsymbol{s}_i \right\rangle}{3\langle \boldsymbol{n} \rangle} \quad if \quad \langle \boldsymbol{M} \rangle = 0$$

ρ = integrated spectra
z = spin quantization axis, // k<sub>photon</sub>

Unlike XMCD or XMLD SP-PES does not require FM or AFM order



#### $Ca_2RuO_4$ : an AFM ( $T_N$ =110K) Mott insulator



#### SP-ARPES - $Sr_2RuO_4$ : a spin-triplet superconductor?



#### SP-ARPES - Sr<sub>2</sub>RuO<sub>4</sub>: a spin-triplet superconductor?



It is not possible to associate a well-defined spin value to a state at the Fermi surface. Cooper pairs cannot be simply described as "singlets" or "triplets"

C.N. Veenstra et al., PRL (2014)

#### Strong SOI in 5d TM compounds: the iridates

